

Processive Sugar Transferase

Field of the invention

The invention relates to the use of processive UDP-sugar: 1,2-diacylglycerol-3- β -, UDP-sugar: 3- β -1,3'-phospho-*sn*-glycerol-1',2'-diacyl-*sn*-glycerol)- and UDP-sugar: 3-[O- β -D-glucopyranosyl]-*sn*-glycerol-1,3'-phospho-1',2'-diacyl-*sn*-glycerol-D-sugar transferases and similar proteins as well as the corresponding coding nucleic acids for the manipulation of the contents and/or the structure of glycosyldiacylglycerols and/or the synthetic secondary products thereof, as well as other substrates which are glycosylated by these enzymes, in transgenic cells and/or organisms.

Glycosyldiacylglycerols were produced enzymatically by means of a sugar transferase (glycosyl transferase). For this purpose, the gene coding for a UDP-sugar transferase was isolated from genomic DNA of *Bacillus subtilis* and *Staphylococcus aureus*, and cloned into, and expressed in, *E. coli*. The activity of the enzymes was confirmed by means of specific *in vitro*-enzyme assays. The products were also detected and identified in lipid extracts of transgenic *E. coli* cells. The products are various novel glycolipids having different number of glucose residues (maximum of 4) linked via a $\beta(1\rightarrow6)$ glycosidic bond, and utilizing diacylglycerol (DAG) or phosphatidylglycerol (PG) as the primary acceptor.

In addition, these novel glycolipids comprise two differently structured novel phosphoglycolipids (**PL1** and **PL2**) with a different number of glucose residues (maximum of two) which in the case of (**PL2**) are also linked via a $\beta(1\rightarrow6)$ glycosidic bond and utilize phosphatidylglycerol as the acceptor (i.e. both diastereomers, i.e. with respect to the configuration of the non-acylated glycerol residue). The glycosyl residues may further be acylated in position 6''' of the terminal glucose:

- 1) MGlcD: 3-[O- β -D-glucopyranosyl]-1,2-diacylglycerol (*Staphylococcus aureus* ypfP)
- 2) DGlcD: 3-[O- β -D-glucopyranosyl-(1 \rightarrow 6)-O- β -D-glucopyranosyl]-1,2-diacylglycerol
- 3) TGlcD: 3-[O- β -D-glucopyranosyl-(1 \rightarrow 6)-O- β -D-glucopyranosyl-(1 \rightarrow 6)-O- β -D-glucopyranosyl]-1,2-diacylglycerol

4) TeGlcD: 3-[O-β-D-glucopyranosyl-(1→6)-O-β-D-glucopyranosyl-(1→6)-O-β-D-glucopyranosyl-(1→6)-O-β-D-glucopyranosyl]-1,2-diacylglycerol

5) Phospholipid 1: 3-[O-β-D-glucopyranosyl]-*sn*-glycerol-1,3'-phospho-1',2'-diacyl-*sn*-glycerol)

6) Phospholipid 2: {3-[O-(6'''-O-acyl)-β-D-glucopyranosyl-(1'''→6'')-O-β-D-glucopyranosyl]- 2-acyl-*sn*-glycerol-1,3'-phospho-1',2'-diacyl-*sn*-glycerol}

Note: The numbering of the glycerol residues I (Gro^I) and II (Gro^{II}) corresponds herein to the numbering 1-3 and 1'-3', respectively, i.e. Gro^I is "left-hand" and Gro^{II} is "right-hand" in accordance with Figure 13.

Surprisingly, the enzymes act in a processive manner, i.e. all detected novel glycolipids are formed by successive addition of UDP-glucose to the respective preceding product of the enzymes. Further, alkyl-β-D-glucosides, ceramides (both enzymes), sterols and sterol glucosides (only the enzyme of *S. aureus*) are used as acceptors for a further glucosylation reaction.

Detailed Description State of the art

Glyceroglycolipids represent a group of membrane components which are very heterogeneous with respect to their structure. They are found in bacteria (Kates, 1990), plants and in very low amounts also in animals. Many structures especially of bacterial glycolipids have already been described many years ago (Kates, 1999), however, none of the genes synthesizing these glycolipids have been cloned, so that these substances can be obtained from the corresponding organisms only in analytical amounts. Only at the beginning of 1997 was the first publication issued, wherein the cloning and expression of a plant galactose : 1,2-diacylglycerol galactosyl transferase is described (Shimojina et al., 1997). However, this enzyme is no "processive" glycosyl transferase.

Database searches in the "U.S. Patent Database" revealed that two further patents relating to glycosyl transferases exist: Patent No. 5 545 554: Glycosyl transferases for biosynthesis of oligosaccharides, and genes and encoding them, and Patent No. 5 641 668: Proteins having glycosyl transferase activity. It appears that the first-mentioned patent only relates to glycosyl transferases which synthesize oligosaccharide, so that this patent is not relevant with respect to the enzyme, viz a lipid glycosyl transferase, described herein. The second-mentioned patent relates to glycosyl transferases in general, in view of which the processive enzyme described in this specification is novel.

Brief Description of the Figures

Industrial Applicability

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Glycosyl diacylglycerols are naturally occurring compounds found in plants, animals and bacteria. However, an inexpensive, large-scale production of these compounds was not possible so far, since corresponding genes were not yet cloned. Glycosyl diacylglycerols can be used in a variety of applications, depending on the number of sugar residues and the structure of the fatty acids.

When esterified with usual C18 unsaturated fatty acids, diglucosyl diacylglycerols have emulsifier properties which are useful in food industrial applications (in mayonnaise, margarine, ice cream, confectionery etc.).

In the presence of highly unsaturated fatty acids, glycolipids may be introduced into polymers, which then obtain new characteristics and surfaces. Finally, glycosyl diacylglycerols may obtain detergent characteristics, when the fatty acid chain length is drastically shortened. This would already now be possible in transgenic rape seed with predominant lauric acid. Such detergents could be produced in large amounts in an inexpensive manner, and such detergents would be biologically degradable.

The phospholipids which are glucosylated by the enzyme of *S. aureus* receive new physico-chemical characteristics due to the charge of the phosphate residue between the two glycerol residues on the one hand, and on the other hand due to acylation of the sugar residue(s). Thus, by use of the described processive sugar transferases, not only neutral lipids, but also charged

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glycolipids can be specifically produced and varied. Thus, a further class of charged glycolipids are developed via the sugar transferases.

In the production of plant oils from oil seeds, a lecithin fraction is obtained, wherein phospholipids and glycolipids are accumulated. By over-expressing the genes disclosed in this specification in these plants, a variety of glycolipids (glucosyl diacylglycerols, steryl glucoside, glucocerebroside and other lipids described herein) could be concentrated, with a favourable effect on the baking properties of bakery products, to which the lecithin fraction is added.

In addition, the phospholipids glucosylated by the *S. aureus* enzyme receive further physico-chemical properties due to the charge of the phosphate residue.

This invention, therefore, relates to a process for the production of glycolipids in transgenic cells and/or organisms, comprising the following steps:

- transfer of a nucleic acid molecule that codes for a protein having the biological activity of a processive diacylglycerol glycosyltransferase to the cells or organism,
- expression of the protein having a biological activity of a processive diacylglycerol glycosyltransferase under suitable regulatory sequences in the cells or the organism, and
- if desired, recovery of the glycolipids synthesized by the biological activity of a processive diacylglycerol glycosyltransferase from the cells or the organism.

In a preferred embodiment of the invention, the nucleic acid molecule codes for a protein having the biological activity of a processive diacylglycerol glycosyltransferase from *Bacillus subtilis* or *Staphylococcus aureus*.

The transgenic cells may be any cells that are useful for the production of the new glycolipids, preferably the cells are plant, yeast or bacteria cells. The transformed organism is preferably a plant, yeast or bacterium.

As mentioned above and as will be clear from the following description, the glycolipids produced by the process of the invention are preferably glucosyl diacylglycerols and/or

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phosphoglycolipids. More preferably, the glycolipids are monoglycosyldiacylglycerol, diglycosyldiacylglycerol, triglycosyl diacylglycerol, tetraglycosyldiacylglycerol, glycosyl ceramide, diglycosyl ceramide, steryl glycoside, steryl diglycoside, glycosyl phosphatidylglycerol, and/or diglycosyl phosphatidylglycerol. Most preferably, the glycolipids are monoglucosyldiacylglycerol, diglucosyldiacylglycerol, triglucosyldiacylglycerol, tetraglucosyldiacylglycerol, glucosyl ceramide, diglucosyl ceramide, steryl glucoside, steryl diglucoside, glucosyl phosphatidylglycerol, and/or diglucosylphosphatidylglycerol.

The invention is also directed to the use of a nucleic acid molecule coding for a protein having the biological activity of a processive diacylglycerol glycosyltransferase or of a proteins having the biological activity of a processive diacylglycerol glycosyltransferase for processive glycosylation, in particular for production of glycolipids. Processive glycosylation, in particular the production of glycolipids, may take place *in vivo* or *in vitro*.

Further, the invention is directed to tetraglucosyldiacylglycerol, synthesized and described herein for the first time. The same applies to glucosylphosphatidylglycerol and diglucosylphosphatidylglycerol.

The invention is further directed to the use of the glycolipids produced by processive glycosylation according to the invention in the food industry, as an emulsifier or as a detergent.

A processive glycosyl transferase, as described herein, catalyzes the successive transfer of one or more hexose residues to an acceptor molecule. In particular the enzyme catalyzes at least one of the following reactions:

- a) addition of hexose $\beta(1 \rightarrow 6)$ to diacylglycerol,
- b) addition of hexose $\beta(1 \rightarrow 6)$ to a MHexD,
- c) addition of hexose $\beta(1 \rightarrow 6)$ to a DHexD,
- d) addition of hexose $\beta(1 \rightarrow 6)$ to a THexD,
- e) addition of hexose $\beta(1 \rightarrow 6)$ to a TeHexD,
- f) addition of hexose β to a ceramide,
- g) addition of hexose $\beta(1 \rightarrow 6)$ to a monohexosyl ceramide,

- h) addition of hexose β to a sterol,
- i) addition of hexose $\beta(1\rightarrow6)$ to a steryl glucoside,
- j) addition of hexose in β -glycosidic linkage to the primary hydroxyl group of phosphatidylglycerol,
- k) addition of hexose $\beta(1\rightarrow6)$ to the first hexose of phosphatidylglycerol- β -D- glucoside.

In particular the glycosyl transferase catalyzes the successive transfer of one or more hexose residues to at least one acceptor molecule for synthesis of glycolipids, in particular phosphoglycolipids, in particular catalyzing one of the following reactions:

- a) addition of Glc $\beta(1\rightarrow6)$ to a diacylglycerol,
- b) addition of Glc $\beta(1\rightarrow6)$ to a MGlcD,
- c) addition of Glc $\beta(1\rightarrow6)$ to a DGlcD,
- d) addition of Glc $\beta(1\rightarrow6)$ to a TGlcD,
- e) addition of Glc $\beta(1\rightarrow6)$ to a TeGlcD,
- f) addition of Glc β to a ceramide,
- g) addition of Glc $\beta(1\rightarrow6)$ to a monoglucosyl ceramide,
- h) addition of Glc β to a sterol,
- i) addition of Glc $\beta(1\rightarrow6)$ to a steryl glucoside,
- j) addition of Glc in β -(1 \rightarrow 6)-glycosidic linkage to the primary hydroxyl group of phosphatidylglycerol,
- k) addition of Glc $\beta(1\rightarrow6)$ to the first Glc of phosphatidylglycerol- β -D-glucoside.

The invention also relates to the DNA sequences coding for an protein having enzyme activity of a processive glycosyl transferase from *Bacillus subtilis* and/or *Staphylococcus aureus*.

Further the invention is directed to DNA sequences coding for a protein which shows at least 50 %, preferably at least 70 %, more preferably at least 90 %, and most preferably at least 95 % identity with the deduced protein of ypfP (Clustal X). More particular, the DNA sequence codes for a protein having more than 5 amino acids within the amino acid sequence EHQPDIH which are identical with the amino acid sequence of the proteins from *B. subtilis* and/or *S. aureus*, preferably having more than 6 amino acids within the amino acid sequence QVVVVCGKN or the amino acid sequence DCMITKPG which are identical with the amino acid sequence of the proteins from *B. subtilis* and/or *S. aureus*. Most preferably, the DNA

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sequence codes for a protein the amino acid sequence of which comprises the amino acid sequence MITKPGGITxTE (wherein x is any amino acid), or the amino acid sequence VKxTGIPI (wherein x is any AA) or the amino acid sequence of which comprises more than 5 amino acids within the sequence ZPDIIIxxxP (wherein Z represents the amino acid Q or K and x is any amino acid) which are identical to the sequence found in *Bacillus subtilis* and/or *Staphylococcus aureus*.

The invention also relates to the use of a processive glycosyltransferase for biosynthetic production of glycolipids having the following structure:

- a) β -D-Glcp-(1 \rightarrow 6)- β -D-Glcp-(1 \rightarrow 6)- β -D-Glcp-(1 \rightarrow 6)- β -D-Glcp-(1 \rightarrow 6)-Gro,
- b) 3-[O- β -D-Glucopyranosyl]-phosphatidylglycerol (PL1), or
- c) {3-[O-(6"-O-acyl)- β -D-glucopyranosyl-(1" \rightarrow 6")-O- β -D-glucopyranosyl]-2-acyl-phosphatidylglycerol} (PL2).

Finally, the invention is directed to secondary products which are produced by biosynthetically and gentechnically engineered microorganisms and/or plants using a processive glycosyl transferase by further conversion of the products produced by the action of the processive sugar transferases, in particular by addition of a fatty acid to the position 6" of the terminal hexose in {3-[O- β -D-glucopyranosyl-(1" \rightarrow 6")-O- β -D-glucopyranosyl]-2-acyl-phosphatidylglycerol}.

1. Isolation and cloning of *ypfP*

The *ypfP* gene was isolated from *B. subtilis*, the gene being described in the SubtiList Database as an open reading frame of unknown function (accession number P54166). The other gene that was isolated and cloned was a sequence from *Staphylococcus aureus* (accession number Y14370) described as an open reading frame of unknown function.

For DNA isolation, restriction analysis and ligation, standard techniques were used (Sambrook et al., 1989). Genomic DNA from *Bacillus subtilis* 019 was isolated according to Cutting et al., 1990. Genomic DNA of *S. aureus* was provided by Prof. Dr. Witte, (Robert

Koch-Institute, Postfach 650280, 13302 Berlin). Restriction endonucleases and DNA-modifying enzymes were purchased from New England Biolabs and Boehringer Mannheim, and used as recommended by the suppliers.

E. coli XL1 Blue (MRF') (Stragene), *E. coli* BL21 (DE3) (Novagen) and *Bacillus subtilis* 019 were grown at 37°C in a Luria Broth (LB) (Sambrook et al., 1989). For plasmid-bearing *E. coli* strains, the antibiotics ampicillin (100 µg ml⁻¹) and kanamycin (30 µg ml⁻¹) were included in the medium. The vectors pUC18 (Yanish-Perron et al., 1985) and pET24c(+) and pET24d(+)(Novagen) were used as cloning vectors. The *yfpP* genes were isolated from genomic DNA of *B. subtilis* and *S. aureus* by PCR. For this purpose the specific primers PJ1 (5'-CCGAGCTCC CATATGAATACCAATAAAAGAG 3') and PJ2 (5' TCCGGATCC TTACGATAGCACTTTGGC 3') for *B. subtilis yfpP* and the primers PJ10 5' TTCC ATGGTTACTCAAAATAAAAAGATATTG 3' and PJ11 5' TTTGGATCCTTATTTAACGAAGAATCTTGCATATAA 3' for the *S. aureus* gene (*say*) were used, the underlined part of which annealed to the 5' and 3' end of the *yfpP/say* genes. The following amplification program was used: 10 min at 94°C; 30 cycles of 0.5 min at 55°C and 60°C for *S. aureus yfpP*, respectively, 2 min at 72°C, 1 min at 94°C; one cycle of 10 min at 72°C. *Pwo*-polymerase (Boehringer) was used for the amplification of the 1170 bp product of the genomic DNA of *B. subtilis*, *Pfu*-polymerase (Stratagene) was used for the amplification of the 1190 bp product from *S. aureus* genomic DNA. The amplified genes were cloned into *Sma*I-linearized pUC18 vector, resulting in *yfpP3* and *psay1*. For construction of the expression vectors p*EyfpP* 24 and p*Esay*24, the *yfpP* fragments were released by *Bam*HI and *Nde*I and *Nco*I digestion, respectively, from *yfpP3* and *psay1*, and inserted into *Bam*HI-, *Nde*I- and *Nco*I-linearized pET24c(+) and pET24d(+), respectively. *E. coli* XL1 Blue (MRF') was transformed with *yfpP3* and *psay1* and *E. coli* BL21 (DE3) was transformed with p*EyfpP*24 and p*Esay*24. Correct in-frame cloning was confirmed by sequencing. One strand of the DNA of *yfpP3* and *psay1* was sequenced using the dideoxy method (automatic sequencer 373A and 377, Applied Biosystems). For computer analysis of the sequences, Clone manager for Windows 4.1 (Scientific and Educational Software) was used. Database searches were performed using the BLAST algorithm (Altschul et al., 1990). Sequence alignments were performed using Clustal X (Higgins and Sharp, 1988).

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2. Expression of the *ypfP/say*-genes

For expression of the genes, *ypfP* was cloned into pET24c(+) and pET24d(+), respectively, and *E. coli* BL21 (DE3) was transformed with the resulting constructs pE*ypfP*24 and pE*say*24. Pre-cultures of *E. coli* BL21 (DE3), *E. coli* BL21 (DE3) pE*ypfP*24 and *E. coli* BL21 (DE3) pE*say*24 were grown overnight at 37°C, and expression cultures were started at an optical density (O.D.)₅₈₀ of 0.05. Induction was performed by adding 0.4 mM IPTG at an optical density of 0.8 and further incubation for 2 h at 37°C. All subsequent steps were carried out at 4°C. Cells were collected by centrifugation (15 min, 5000x g). The cell pellet was re-suspended in a buffer 1 (50 mM Tris-HCl, pH 8.0; 20% (v/v) glycerol) (4% of the volume of the expression culture). The cells were frozen and sonicated after thawing (3x 40s; Braun, Labsonic 2000). Inclusion bodies were collected by centrifugation (15 min, 4000x g) and the supernatant was then divided into the membrane fraction and the soluble supernatant (*B. subtilis ypfP*) by ultrasonification (1 h, 147000x g).

The inclusion bodies, the membrane fraction and the soluble supernatant were separated in an SDS-PAGE. The SDS-PAGE was carried out as described by Laemmli, 1970, and the gels were stained with Coomassie brilliant blue R250 (Serva).

By SDS-PAGE analysis, over-expression of a protein having an apparent molecular mass of 44 kDa could be identified in the membrane fraction and the inclusion body fraction. The molecular weight corresponds to the calculated mass of 43.6 kDa or 44.7 kDa, respectively, for YpfP. This protein was not present in the soluble fraction and in untransformed *E. coli*.

3. Lipid extraction and analysis

Expression cultures of *E. coli* BL21 (DE3) pE*ypfP*24 and pE*say* 24 and cultures of the late logarithmic growth stage of *Bacillus subtilis* 019 were harvested by centrifugation (15 min, 5000x g), and the sedimented cells boiled for 10 min in water. Lipid extraction was performed as described by Linscheid et al., 1997. For separation of individual lipids by preparative chromatography, the lipids were subjected to thin-layer chromatography in the following solvent systems: (1) chloroform/methanol/H₂O (70:30:4. v/v/v) for separation of MGlcD, DGlcD, TGlcD and TeGlcD from phospholipids; (2) diethyl ether/petroleum ether (2:1, v/v) for separation of acetylated DGlcD from non-acetylated DGlcD; (3) diethyl ether/petroleum

for separation of acetylated DGlcD from non-acetylated DGlcD; (3) diethyl ether/petroleum ether (4:1, v/v) for separation of acetylated TGlcD from non-acetylated TGlcD; (4) chloroform/acetone (9:1, v/v) for separation of acetylated TeGlcD from non-acetylated TeGlcD.

Separation of the two acetylated phosphoglucolipids **PL1** and **PL2** was performed in the solvent chloroform/methanol (80:20, v/v). Then both acetylated lipids were extracted from the silica gel and re-suspended in chloroform. The lipids were methylated by addition of diazomethane and subsequently separated in the solvent toluene/methanol (9:1, v/v).

Acetylation of the glycolipids was performed as described by Tulloch et al., 1973. Synthesis of the fatty acid methyl esters from DGlcD with sodium methylate was performed according to Roughan and Beevers, 1981. Release of the fatty acid from the *sn* 1 position of DGlcD was achieved by incubation with *Rhizopus* Lipase (Boehringer) according to suppliers' protocol. Incubation with Cerebrosidase (provided by Prof. Dr. Sandhoff, University Bonn) was performed as described by Vaccaro et al., 1993.

The lipid extracts of *E. coli* BL21 (DE3) pEyp/P24 and pEsay24 showed various new glycolipids, which could not be detected in the wild-type (Fig. 1). These glycolipids reacted with a sugar-specific spray reagent, but they were ninhydrin and phosphate negative (the native **PL1** and **PL2** were phosphate positive). One of the glycolipids co-migrated with a diglucosyl diacylglycerol (DGlcD) standard of *B. cereus*. The different glycolipids were purified and acetylated. The glycolipid band with the polarity of DGlcD also co-migrated after acetylation with the acetylated DGlcD standard of *B. cereus*.

4. Analysis of the new glycolipids by MS and NMR

Mass spectrometric (MS) and nuclear magnetic resonance spectroscopic (NMR) analysis of the new glycolipids was exclusively performed with the per-*O*-acetylated derivatives (**1**, **2**, **3**, **4**.) and the phosphomethyl esters (**PL1**, **PL2**) of the glycolipids, respectively.

4.1. Mass spectrometric analysis (CI-MS and MALDI-MS)

4.4.1. EI-MS and CI-MS (DIP-mode)

Mass spectrometric analysis of the neutral glycolipids was carried out with a Hewlett Packard mass spectrometer (Model 5989) using the direct insert probe (DIP) mode. The sample was evaporated from 80°C to 325°C at a rate of 30°C/min.

While all per-*O*-acetylated di-(2), tri- (3) and tetrahexosyl-(4) diacylglycerolipids could be analyzed by MS analysis in the DIP mode directly, the two phospholipids PL1 and PL2 could not be analyzed by this technique. Due to the high polarity and the complexity of the molecule, the phospholipids were, therefore, dephosphorylated with hydrogen fluoride (48% HF, 4°C, 20h) prior to MS analysis, the dephosphorylated fragment was per-*O*-acetylated and only after this treatment analyzed by mass spectrometry (DIP mode). Electron impact spectra (EI-MS) were recorded at 70 eV and chemical ionization spectra (CI MS) were obtained using ammonia (0.5 torr).

In the DIP MS analysis all per-*O*-acetylated di-(2), tri- (3) and tetrahexosyl-(4) diacylglycerolipids in the EI mode showed characteristic fragments for terminal mono-hexosyl ($m/z = 331$) and di-hexosyl ($m/z = 619$) residues and differed from each other in the evaporation rate maximum (9.5 min, 2, 10.6 min, 3 and 12.0 min, 4). The disaccharid 2 showed in the CI MS a pseudomolecular ion $[M + NH_4]^+$ at $m/z = 1202$, wherein hexadecanoyl (16:0) and hexadecenoyl (16:1) could be identified as the fatty acid residues. In addition, a second ion $[M + NH_4]^+$ at $m/z = 1230$ was observed, which could be identified as disaccharide with 16:0 and 18:1 (or 18:0 and 16:1) as fatty acids. The amounts of these differently acetylated diglycosyl lipids were present in a relative proportion of 2:3.

The trisaccharide 3 showed the expected pseudomolecular ion $[M + NH_4]^+$ at $m/z = 1490$ and 1516 with the same heterogeneity in its acylation pattern, however in a slightly different proportion (2:1).

The tetrasaccharide 4 showed an evaporation profile with an increased maximum in the evaporation time (12.0 min) in comparison to 2 and 3. Pseudomolecular ions in the CI MS

could not be produced with this compound. The presence of the tetrasaccharide **4** could, therefore, only be indirectly deduced under these conditions from the characteristic fragments of the non-reducing glycosyl residue ($m/z = 331$ and 619 , respectively).

In the DIP MS (CI mode) both **PL1** and **PL2** showed a characteristic biphasic evaporation profile, wherein the first maximum (~ 6 min) from the dephosphorylated and re-acetylated partial structures and the second maximum (~ 11 min) from the diagnostic pyrolysis fragments of the intact, not cleaved by HF hydrolysis phospholipids could be assigned. From this result it had to be noted that the applied reaction time (6 –20h) for removal of the phosphate residue with aqueous (48%) HF was not sufficient, since both phospholipids were cleaved only partially into their dephosphorylated partial fragments. In spite of this limitation, the fatty acid distribution pattern for both of the glycerol residues Gro^I and Gro^{II} could be unambiguously determined.

For **PL1**, the first peak revealed two pseudomolecular ions $[M + NH_4]^+$ at $m/z = 626$ and 654 , which could be assigned to the molecule ions and, thus, to the mono-acetylated glycerol residue Gro^{II} with 16:0 and 16:1 ($M = 608$) and a 16:0 and 18:1 (or 18:0 and 16:1) ($M = 636$), respectively. Thus, this peak contained the expected product of the HF treatment and re-acetylation. In contrast, an analogous reaction product deduced from Gro^I (with Glc as the substituent) was not observed.

The second peak (~ 11.0 min) revealed three fragments, which in contrast to the first peak represented only pyrolysis products of **PL1**, produced during MS analysis, but no intact derivatives. The first ion ($m/z = 447$) was assigned to a fragment deduced from Gro^I, substituted with a peracetylated Glc and with an acetyl residue. The two other fragments ($m/z = 549$ and 577) originate from Gro^{II}. They carried a diacylglycerol, wherein each a palmitic acid (16:0) and a palmitoleic acid (16:1) were esterified ($m/z = 549$), and a second fragment ($m/z = 577$) with each a palmitic acid (16:0) and an oleic acid (18:1) (alternatively 16:1 and 18:0). Already from this fragmentation pattern it could be concluded that **PL1** is an “asymmetrically acylated” phospholipid, because Gro^{II} carries both fatty acids, and Gro^I in the native phospholipid with a free hydroxyl group and a glucose appears to be relatively hydrophilic.

The dephosphorylated and per-*O*-acetylated **PL2** also showed a biphasic evaporation profile in the DIP MS analysis. The first peak had an evaporation maximum (~ 6 min) and a fragmentation pattern identical with **PL1**, indicating that **PL2** is identical with **PL1** with respect to the fatty acid substitution pattern in Gro^{II}. In contrast, the second peak (11.0 min) revealed four fragmentations. The first couple ($m/z = 550$ and 577) was identified as a pyrolysis fragment deduced from Gro^{II} with 16:0 and 18:1 (16:1 and 18:0, respectively) and was therefore analogous to **PL1**. The second couple ($m/z = 1202.9$ and 1231.1) could be identified as $[M + NH_4]^+$ ion of an intact derivative, produced from **PL2** after dephosphorylation and re-acetylation. This **PL2** derivative is a diglucosyl diacylglycerol being further esterified with two fatty acids 16:0 and 16:1 ($M = 1185$) and 16:0 and 18:1 ($M = 1203$) respectively, as well as an acetyl residue (on the initial phosphate position). Although it was possible to assign the fatty acid substitution pattern to the two glycerol residue Gro^I and Gro^{II} by these analyses, the exact substitution pattern of the fatty acids could only be determined by NMR analysis (see below). Due to thermal instability of the molecules, the DIP MS analyses of the two intact phospholipid derivatives **PL1** and **PL2** could not be completely analyzed and were therefore further examined by means of MALDI-TOF-MS analysis.

4.1.2. MALDI-TOF-MS

MALDI-TOF-MS analyses were carried out on a Bruker Reflex II spectrometer in the reflector mode at an acceleration potential of 20 kV by means of the “delayed ion extraction” in the positive mode. The per-*O*-acetylated and phosphomethylated samples of **PL1** and **PL2** were re-suspended in chloroform (10 µg/ml) and 2 µl solution thereof were mixed with 2 µl of a matrix solution (0.5 M 2,4,6-trihydroxyacetophenone; Aldrich, Steinheim). An aliquot of this mixture (0.5 µl) was applied to a metal support, dried with warm air and immediately thereafter placed into the spectrometer. Calibration of the spectra was performed using an internal standard (Angiotensin). All mass data apply exclusively to the monoisotopic mass of the molecules.

In each case, two derivatives of both phospholipids were analyzed by MALDI-TOF-MS: the free phosphoric acid derivatives and the phosphomethyl esters (**PL1** and **PL2**, see Fig. 2). The free phosphoric acid derivative of **PL1** showed prior to esterification of the phosphate residue (diazomethane) in the positive reflector mode a pseudomolecular ion $[M-H+Na]^+$ at $m/z =$

1116,48, corresponding to the calculated formula $C_{54}H_{95}O_{20}P$ ($M = 1094,56$). The non-esterified phosphoric acid derivative of **PL2** showed under the same conditions a pseudomolecular ion $[M-H+Na]^+$ at $m/z = 1796,43$, corresponding to the formula $C_{94}H_{167}O_{28}P$ ($M = 1775,06$) and thus carrying in addition a hexose and a fatty acid (16:0 and/or 18:1) in comparison to **PL1**.

Note: The mass of the MALDI-TOF-MS analyses given herein in all cases only relate to the smallest monoisotopic pseudomolecule ion or mass fragment. That means, in all mass data only the smallest fatty acid (16:0) was considered (see Fig. 2). All pseudomolecular ions presented herein showed per fatty acid always a heterogeneity resulting from the exchange of 16:0, 16:1, 18:0 and 18:1, this heterogeneity influencing all mass spectra (DIP and MALDI), but is not considered in the mass data provided herein.

PL1 showed after treatment with diazomethane a pseudomolecular ion $[M-H+Na]^+$ at $m/z = 1130,69$ which corresponds to the formula $C_{55}H_{97}O_{20}P$ ($M = 1108,57$) and thus contains only one additional methyl group ($\Delta m/z = 14$) in comparison to the free acid. **PL2** showed under the same conditions a pseudomolecular ion $[M-H+Na]^+$ at $m/z = 1811,42$, which corresponds to the formula $C_{95}H_{169}O_{28}P$ ($M = 1789,07$) and also contains only one additional methyl group ($\Delta m/z = 14$) in comparison with the free acid. Thus, not only the preceding MS analyses (DIP MS) were confirmed, but it was also unambiguously demonstrated that both **PL1** and **PL2** represent phosphodiester, which are likely to be substituted with two glycerol residues. In both cases only one methyl group was introduced by a diazomethane treatment and transformation into the corresponding methyl ester.

4.2. Proton Nuclear Magnetic Spectroscopic Analysis (1H -NMR)

The per-*O*-acetylated and purified samples (**2–4**, 30-200 μg) were dissolved in 100 μl $CDCl_3$ (99.96% Cambridge Isotope Laboratories, Andover, MA, USA), and transferred into special capillary NMR microtubes (2.5 mm OD, Wilmad, Buena, NJ, U.S.A.). The proton spectra (1H -NMR) were recorded on a 600 MHz Spectrometer (Bruker Avance DRX 600), equipped with a special microprobe head (PH TXI 600SB). The samples were measured at 300K with reference to internal trimethylsilane (TMS, $\delta_H = 0.000$ ppm). One- and two-dimensional

homonuclear spectra (^1H , ^1H COSY, ROESY, and relayed COSY) were performed using standard Bruker software (XWINNMR, Version 1.3).

The one-dimensional (1D) ^1H -NMR spectra (600 MHz, microprobe head) of the di- (2, ≈ 200 μg), tri- (3, ≈ 200 μg), and tetrahexosyl diacylglycerolipids (4, ≈ 50 μg) of compounds 2, 3, and 4 are shown in Figures 3a –c and the results are set forth in Table 1 (Annex).

Assignment of the signals was carried out by 1D and two-dimensional (2D) proton nuclear magnetic resonance spectrometry (^1H , ^1H COSY, relayed ^1H , ^1H COSY, ROESY) in comparison with the structurally related β -gentiobiose octaacetate (1) which served as a reference substance for unambiguous assignment and which is therefore also included in Table I. The β -anomeric configuration of all hexoses in the substance results from the coupling constants $J_{1,2}$ being between 7.6 and 8 Hz for all glucoses. The other coupling constants of the pyranosidic ring protons H-2, H-3, and H-4 and H-5 ($J_{2,3}$, $J_{3,4}$, $J_{4,5}$) were all larger than 9.5 Hz, indicating glucopyranose. The chemical shift of the methylene protons (H-6a and H-6b) as well as their coupling constants ($J_{6a, 6b}$) in the terminal Glc residue (**A**) were found to be identical in all oligosaccharides (4.062 ± 0.005 ppm for H-6a and 4.205 ± 0.005 ppm for H-6b) as compared with those of H-6a,6b in residue **A** of the β -gentiobiose octaacetate, thus allowing the assignment of the spin systems of all terminal Glc residues **A** in the oligosaccharides 2, 3, and 4 on the one hand, but also in PL1 and PL2.

The $\beta(1 \rightarrow 6)$ glycosidic bond could be determined by means of the shift towards a higher field of the (overlapping) signals of H-6a and H-6b in residues **B**, **C** and **D** (3.855 ± 0.05 ppm) since these signals clearly differed from the non-substituted methylene signals of the terminal H-6a,6b (**A**). This fact clearly indicates that all Glc residues of the compounds 2, 3, and 4 are identically, i.e. $\beta(1 \rightarrow 6)$ glycosidically interlinked. This observation could be confirmed by means of a two-dimensional spectrum (^1H , ^1H COSY, Figure 4, bottom) and a nuclear Overhauser spectrum (rotating-frame NOE spectroscopy, ROESY, Figure 4, above) of trisaccharide 3. A ROESY spectrum showed (indicated) cross-peaks of the anomeric H-1 protons H-1^A, H-1^B, and H-1^C which could be observed between H-1^A/H-6a,6b^B, H-1^B/H-6a,6b^C, and H-1^C/H-3a,3b^{GrO} (Fig. 5), allowing an unambiguous assignment of the three spin systems to each of the glucosyl residues **A**, **B** and **C**.

In addition to the glycosyl residues in all ^1H -NMR spectra, signals of the glycerol moiety (H-1a, 1b^{Gro}, H-2^{Gro}, and H-3a,3b^{Gro}) could also be identified (Fig. 3, Table 1). The fatty acids showed the expected methylene ($-\text{CH}_2-$, 1.185 ppm) and methyl protons ($-\text{CH}_3$, 0.812 ppm). Finally, signals from olefinic protons ($-\text{CH}=\text{CH}-$, ≈ 5.27 ppm) could also be found in all glycolipids **2-4**, **PL1** and **PL2**, which from the MS spectra could be assigned to the unsaturated fatty acids 16:1 and/or 18:1.

In accordance with the non-phosphorylated compounds **1 – 4**, NMR analyses were carried out only with the per-*O*-acetylated mono methyl ester derivatives. Samples (0.1 – 0.2 mg) were dissolved in 500 μl CDCl_3 (99.96 % Cambridge Isotope Laboratories, Andover, MA, U.S.A.) and measured in 5 mm NMR tubes (Ultra Precision NMR sample tubes, Isocom, Landshut) at 300 K. Proton and Phosphorous-31 spectra (^1H - and ^{31}P NMR) were recorded with a 600 MHz spectrometer (Bruker Avance DRX 600) equipped with an inverse probe head (5 mm TXI 13C), and the carbon 13 (^{13}C NMR) spectra were recorded with 360 MHz Bruker AM spectrometer (5mm dual probe head) at 90,6 MHz. The chemical shift was measured with reference to internal tetramethylsilane (TMS, $\delta_{\text{H}} = 0.000$ ppm) and chloroform (CHCl_3 , $\delta_{\text{C}} = 77.00$ ppm), respectively. ^{31}P NMR spectra were recorded at 242.9 MHz and calibrated with reference to an external standard (85% $\text{H}_3\text{PO}_4 = 0.0$ ppm). One (1D) and two-dimensional (2D) homonuclear spectra (^1H , ^1H COSY, NOESY, and relayed COSY) and heteronuclear spectra [^1H , ^{13}C and ^1H , ^{31}P HMQC (heteronuclear multiple quantum coherence) as well as ^1H , ^{13}C HMBC (hetero multiple bond correlation)] were recorded with a standard Bruker software (XWINNMR, Version 1.3).

In the ^1H -NMR spectrum (Fig. 6 and Table 2), **PL1** showed, in accordance with MGlcD: {3-[*O*- β -D-glucopyranosyl]-1,2 diacylglycerol}, characteristic signals corresponding to a β -glycosidically bonded Glc residue. Surprisingly, the anomeric proton H-1 was cleaved into a couple of signals (H-1 and H-1') with a similar intensity (H-1, 4.461 ppm, $J_{1,2}$ 7.9 Hz; H-1', 4.457 ppm, $J_{1,2}$ 7.9 Hz. While the other protons (H-2, H-3, H-4, H-5, H-6a,b) showed identical chemical shift and coupling constants as compared with MlcD, the protons of H-3a^{Gro I} and H-3b^{Gro I} (3.62 and 3.88 ppm) on the one hand and H-1a^{Gro II} and H-1b^{Gro II} (4.11 and 4.31 ppm) on the other hand were split: fine resolution of the two other methylene proton signals of the glycerol residues I and II (H-1a,b^{Gro I} and H-3a,b^{Gro II}; $\sim 4.08 - 4.14$ ppm) could, however, not be observed. Further, we observed two singular methine protons for H-2^{Gro I}

(ddd, 5.082 ppm, 5.3 Hz) and H-2^{Gro II} (ddd, 5.168 ppm, 5.3 Hz) as to be expected for a diglyceride.

Further characteristic signals were of the methyl group of the phosphate ester, which also exhibited a characteristic split doublet (POCH₃, 3.828 and 3.810 ppm) with a characteristic $J_{H,P}$ coupling of 11.2 Hz. A phosphate monomethylester, **PL1** could be identified as a phosphodiester via the integral of the signal of the phosphomethyl group (3H). This confirmed the results of MS analyses. Finally, 5 OAc signals could be detected (2.026, 2.018, 1.989, 1.954, 1.934 ppm; all s), which led to the conclusion that, besides the four OAc groups of the terminal Glc residue, presumably a fifth OAc group was bound to one of the two glycerol residues. The accurate fatty acid distribution pattern could be partially determined using an HMBC experiment. A fatty acid in position *sn*-1 of glycerol residue II could be assigned via the connectivities of the α -methyl protons of the fatty acids (-O-COCH₂-). However, due to the small amount of substance, the substitution of the second fatty acid could not be determined in the HMBC experiment and could only be investigated based on MS analyses.

In ³¹P NMR (Figure 7 and Table 3), the phosphate signal of **PL1** is split (0.514 and 0.444 ppm) and appears as a singlet in the decoupled spectrum. The ¹H, ³¹P-HMQC experiment (i) showed the expected connectivity with the phosphomethyl ester group (3.828 and 3.810 ppm), and (ii) revealed two methylene protons of glycerol residue I (H-1a,b^{Gro I} and H-3a,b^{Gro II}) (~ 4.08 – 4.14 ppm) to which the phosphomethyl group is bound. Hence, the phosphate substitution could be determined. Thus, based on this experiment, the connection of glycerol residues I and II via a phosphate diester could be proven, which could already be assumed due the presence of phosphomonomethyl ester in NMR analysis and due to characteristic fragments in MS analysis.

The splitting of the signals of H-1^{Glc}, H-3a,b^{Gro I} and H-1a,b^{Gro II} is especially notable. This anomaly in ¹H NMR can be explained by the presence of a pair of diastereomers of **PL1**. By introducing a methyl group, the prochiral phosphate (R-O-PO(OH)-OR') in the middle of the molecule becomes chiral (R-O-PO(OMe)-OR') which results in two diastereomeric phospholipids **PL1** and **PL1'**. A corresponding chirality of the phosphorous atom was already

observed and described for other phospholipids in ^1H -, ^{13}C and ^{31}P -NMR spectra (Bruzik et al., 1983).

Phospholipid 2 (**PL2**) showed the same characteristic splitting of two anomeric protons in ^1H -NMR. In this case, the anomeric protons are of the two glucose residues Glc^{A} and Glc^{B} ($\text{H}-1^{\text{A}}$ and $\text{H}-1'^{\text{A}}$; 4.647 and 4.635 ppm, $J_{1,2}$ 7.7 Hz) and $\text{H}-1^{\text{B}}$ and $\text{H}-1'^{\text{B}}$ (4.533 and 4.524 ppm, $J_{1,2}$ 7.9 Hz). These splittings are characteristic for diastereomeric pairs analogous to **PL1**. Thus, a structural relationship of both phospholipids and, as a consequence, a correlation in the biosynthesis of these phospholipids can be assumed. The ^1H -NMR spectrum of **PL2** showed high similarity with the one of DGlcD (**2**) which made the interpretation of the NMR spectra and thus the structural analysis easier. By comparison of both ^1H -NMR spectra, the substitution of the fourth acyl residue (16:0 or 18:1) in position C-6^A of terminal glucose could be determined. (The third acyl residue is connected with the C2 group of Gro^I).

Further characteristic signals were of the phosphomethyl ester, which also exhibited a split doublet revealing the chirality of the phosphate residue, and thus the diastereomeric nature of the molecule (3.835 and 3.818 ppm, $J_{\text{P,H}}$ 11.2 Hz). In ^{31}P -NMR, the phosphate signal of **PL2** was also split (0.414 and 0.275 ppm) analogous to **PL1**, which is characteristic for the diastereomeric pair **PL2** and **PL2'** (Bruzik et al., 1983).

In conclusion, our MS (DIP and MALDI) and ^1H -, ^{13}C - and ^{31}P -NMR analyses unequivocally identified three neutral and two inogenic glycolipids, which could be identified as di-, tri-, and tetrasaccharide-diacylglycerols **2**, **3**, and **4** with the following structure in the glycosyl moiety (Figure 8):

$\beta\text{-D-Glcp}-(1\rightarrow6)\text{-}\beta\text{-D-Glcp}-(1\rightarrow6)\text{-Gro}$ (**2**) (DGlcD),

$\beta\text{-D-Glcp}-(1\rightarrow6)\text{-}\beta\text{-D-Glcp}-(1\rightarrow6)\text{-}\beta\text{-D-Glcp}-(1\rightarrow6)\text{-Gro}$ (**3**) (TGlcD), and

$\beta\text{-D-Glcp}-(1\rightarrow6)\text{-}\beta\text{-D-Glcp}-(1\rightarrow6)\text{-}\beta\text{-D-Glcp}-(1\rightarrow6)\text{-}\beta\text{-D-Glcp}-(1\rightarrow6)\text{-Gro}$ (**4**) (TeGlcD),

as well as the phospholipid 1:3-[*O*- $\beta\text{-D-glucopyranosyl}$]-*sn*-glycerol-13'-phospho-1',2'-diacyl-*sn*-glycerol (**PL1**) and the phospholipid 2: {3-[*O*-(6'''-*O*-acyl)- $\beta\text{-D-glucopyranosyl}$ -(1''' \rightarrow 6'')-*O*- $\beta\text{-D-glucopyranosyl}$]-2-acyl-*sn*-glycerol-1,3'-phospho-1'2'-diacyl-*sn*-glycerol} (**PL2**) (Figure 9).

5. Enzyme Assay

Standard enzyme assays for determination of the activity of the processive glycosyl transferase were performed in a final volume of 100 μ l, containing buffer 1, 20 μ l *E. coli* BL21 (DE3) pEypfP24 and pEsay24 membrane fraction (20-40 μ g of protein) and 250 000 dpm UDP-[14 C]-glucose (specific activity 10.8 GBq/mmol; 3.85 μ M final concentration). The reaction was carried out for 1 h at 30°C and stopped by the addition of chloroform/methanol (2:1; 2 ml). The organic mixture was washed with 0.7 ml of NaCl solution (0.45% (w/v)) and the resultant subphase recovered. An aliquot of the subphase was subjected to scintillation counting, and after removal of the solvent by evaporation with argon, the remaining part was used for separation by thin-layer chromatography.

Detergents such as octyl- β -D-glucopyranoside (Sigma), decyl- β -D-glucopyranoside (Sigma), SDS, Chaps (Sigma), Tween 20, dodecyl- β -D-maltoside (Sigma) and sodium cholate (Sigma) were added in concentrations according to twice their critical micellar concentration (this applies only to the processive glycosyl transferase from *B. subtilis* ypfP).

Ceramide was added as fluorescent D-erythro-C6-NBD-ceramide (Matreya, INC.), cholesterol was added as [4- 14 C]cholesterol and steryl glucoside was added as 14 C-labelled steryl glycoside (Fig. 10/11). Radioactive products on thin-layer chromatography plates were detected by radio scanning (BAS-1000 Bio Imaging Analyzer, Fuji) (see Figure 10).

Assays with UDP-[14 C]glucose showed the highest incorporation of radioactivity with membrane fractions compared with the soluble and inclusion body fractions of *E. coli* BL21 (DE3) pEypfP24 and pEsay24. Therefore, all subsequent *in vitro* standard assays were carried out with membrane fractions and UDP-[14 C]glucose. The [14 C]-labeled lipophilic products counted for 70 –80 % of the label offered in the assay. Separation by TLC was used to identify lipophilic radioactive products, using a monogalactosyl diacylglycerol (MGD), DGlcD and TGlcD as non-radioactive standards. The highest proportion of radioactivity was found in DGlcD, whereas labeling of MGlcD and TGlcD (TeGlcD only for *B. subtilis*) was low (Fig. 11). Assays with membrane fractions of the untransformed *E. coli* did not show incorporation of radioactivity into lipophilic products. To increase the DAG concentration in the enzyme assay, the effects of several detergents on the enzymatic activity were tested. With the exception of lyso-PC (Sigma) and alkyl- β -D-glucopyranosides, the addition of all above-

mentioned detergents resulted in complete inhibition of enzymatic activity. [^{14}C]-MGlcD and [^{14}C]-DGlcD from assays with transformed *E. coli* were isolated and subjected to various chemical and enzymatic treatments to identify their structure.

The DAG moiety in [^{14}C]-DGlcD was confirmed by treatment with *Rhizopus* lipase. This lipase specifically releases the fatty acid from the *sn* 1-position of the DAG-containing lipid. As expected, the resulting radioactive product co-migrated with a lyso-DGlcD that had been prepared from non-radioactive DGlcD by the same treatment. Incubation of [^{14}C]-DGlcD with sodium methylate resulted in the release of a free fatty acid methyl ester and [^{14}C] glucosyl diacylglycerol, the same products were produced when using non-radioactive DGlcD of known structure. Characterization of the linkage between the first glucose and the DAG was carried out by incubation of the labeled MGlcD with cerebrosidase. This enzyme is specific for the β -glucosidic linkage, but is relatively unspecific for the hydrophobic part of its substrate (Vanderjagt et al., 1994). The incubation of [^{14}C]glucose-labeled MGlcD with cerebrosidase resulted in the release of labeled glucose and unlabeled DAG. The success of the hydrolysis was measured by scintillation counting of the aqueous and organic phase after phase partitioning. 90% of the label was found in the aqueous phase as compared with 15% in the control experiment, in which 85% of the radioactivity was recovered as [^{14}C]MGlcD in the organic phase. These results support the assumption of a β -glucosidic linkage between the first glucose and DAG in MGlcD.

6. Characterization of glycosyltransferase activity

The formation of three different radioactive products in the *in vitro* enzyme assay raises the question whether all of these products are produced by a single enzyme coded by *ypfP* genes. To answer this question, three of the possible sugar acceptors were incubated separately in labeled form with unlabeled UDP-glucose in the presence of the membrane fraction. The sugar acceptors were isolated from previous assays. Assays with radioactive [^{14}C]DAG [^{14}C]MGlcD and [^{14}C]DGlcD were performed by sonification of the radioactive substrates in 0.5 mM lyso-phosphatidylcholine (for [^{14}C]DAG) or in ethanol before adding the membrane fraction, buffer 1 and UDP-glucose (3.6 mM final concentration). The maximum ethanol concentration in assays was 5 % (v/v). After conversion of the substrates, the lipophilic products were separated by TLC and detected by radio scanning (Fig. 12). [^{14}C]DAG was

converted to [^{14}C]DGlcD and [^{14}C]TGlcD, [^{14}C]MGlcD to [^{14}C]DGlcD and [^{14}C]TGlcD and [^{14}C]DGlcD to [^{14}C]TGlcD. Conversion of radioactive labeled DGlcD to TGlcD did not occur any more with the *S. aureus* enzyme. Control experiments using the same substrates and untransformed *E. coli* membrane fractions did not result in any of the mentioned products. The results suggest processivity of the enzyme, whereby the starting reaction can be described as a UDP-glucose: 1,2-diacylglycerol-3- β -D-glycosyltransferase reaction. In subsequent reaction steps, however, the glucose acceptors vary and represent the products of previous additions of β -glucosyl residues.

To exclude a reaction mechanism based on the transfer of glycosyl residues from glycosides to various acceptors, as observed for glycosidases, the enzyme assay was carried out in the presence of a radioactively labeled MGlcD, but in the absence of UDP-glucose. No conversion of a radioactively labeled MGlcD could be observed. Incubation of Ypfp with the glucosidase inhibitor deoxynojirimycin (Alexis Deutschland GmbH) and substance 3 (provided by Dr. Y. Ichikawa) was performed as described by Ichikawa and Igarashi, 1995. These compounds interfere with the transfer of glucose in reactions catalyzed by glucosylhydrolases, but not with the transfer of a sugar nucleotide-dependent glucosyltransferases. None of the inhibitors was able to inhibit the enzyme reaction. Both approaches suggest a transfer of glucose by a sugar nucleotide-dependent reaction. On the other hand, ricinoleic acid and oleic acid were able to inhibit the enzyme, inhibition varying with the concentration in the assay. Additions between 25 and 50 μg in 100 μl assay volume resulted in inhibition of DGlcD and TGlcD formation, the second and third step of the enzyme reaction. In these experiments, MGlcD accumulated in the assay, whereas MGlcD accumulated in normal assays in a low amount. Concentrations above 50 μg in the assay led to a complete inhibition of the enzyme. Hydrolysis experiments with sodium methylate excluded the possibility that ricinoleic acid (=12-D-hydroxy-oleic acid) was glucosylated.

7. Substrate specificity

Substrate specificity was characterized regarding the sugar donor and the sugar acceptor. Apart from UDP-[^{14}C]glucose, UDP-[^{14}C]galactose was also tested, but galactose was not incorporated into lipophilic products. Experiments concerning the sugar acceptor showed that besides DAG, MGlcD and DGlcD also alkyl- β -D-glucopyranosides can serve as acceptor

(this applies only for the *B. subtilis* enzyme). This resulted in products, which tentatively have been identified as alkyl diglucosides. However, the only evidence available so far are the R_f-values of the resulting products and their stability towards alkaline hydrolysis. Neither alkyl- α -D-glucopyranoside nor alkyl- β -D-glucopyranoside could serve as acceptor. The *S. aureus* enzyme could convert sterol, as well as steryl glucoside (Figure 10). This data shows that the YpfP enzymes are less specific concerning the sugar acceptor, but have a higher specificity for the sugar donor UDP-glucose.

General cloning and transformation techniques

The recombinant DNA molecules according to the invention can be produced by standard techniques, as, for example, described in the Laboratory Manual by Sambrook et al., vide supra. Also, production of transgenic cells and organisms can be performed using conventional transformation methods, well-known in the art. This applies to microorganisms and yeast, as well as to plants. For introducing DNA into a plant host cell several techniques are available and the person skilled in the art can easily choose a suitable transformation procedure. These techniques comprise the transformation of plant cells with T-DNA using *Agrobacterium tumefaciens* or *Agrobacterium rhizogenes* as transformation means, fusion of protoplasts, direct gene transfer of isolated DNA into protoplasts, microinjection or electroporation of DNA, introducing DNA via biolistic methods and other procedures. In an alternative embodiment of the invention the nucleic acid molecules of the invention can be introduced into plant cells via viral infection. These techniques are all described in the literature, as are suitable binary vectors and expression vectors.

For constructing the recombinant nucleic acid molecules according to the invention, the skilled person can use any DNA sequence that codes for a protein having the biological activity of a processive diacylglycerol glycosyltransferase, including DNA sequences which hybridize with the DNA sequences disclosed herein. In the context of the invention, the term "hybridization" means a hybridization under conventional conditions, preferably under stringent conditions, as e.g. described in Sambrook et al. (1989), Molecular Cloning: A Laboratory Manual, 2nd edition, Cold Spring Harbor Laboratory Press, Cold Spring Harbor, New York. Nucleic acid molecules that hybridize with the molecules of the invention may isolated e.g.

from genomic or cDNA libraries. Identification and isolation of such nucleic molecules can be carried out using the nucleic acid molecules of the invention or fragments or reverse complements thereof, e.g. by hybridization under standard conditions (Sambrook *et al.*, *supra*). For instance, nucleic acid molecules which display exactly or essentially the nucleic acid sequences of the invention or portions thereof can be used as hybridization probes. Also synthetic fragments, which are synthesized using common synthesis processes and which correspond basically to one of the DNA sequences or nucleic acid molecules of the invention can be used as hybridization probes. When genes are identified and isolated, which hybridize to the DNA sequences of the invention, it is necessary to determine their sequence and the sequence and characteristics of the proteins encoded by them. The man skilled in the art has a variety of biochemical, biotechnological and genetic engineering methods for the characterization of the nucleic acid molecules and the proteins at his disposal. The molecules that hybridize to the nucleic acid molecules of the invention comprise also fragments and (degenerated or allelic) derivatives of the nucleic acid molecules described herein. The terms "derivatives" means in this context that the sequences of such molecules differ from the molecules of the invention (as herein described) in one or more positions and show a high degree of homology to the sequences provided by the invention. Homology means an identity in sequence of at least 40 %, in particular of at least 60 %, preferably of more than 80 % and more preferably of more than 90%. The deviations from the sequences of the invention may be generated by deletion, addition, substitution, insertion or recombination.

Abbreviations

AA	Amino acid
DAG	Diacylglycerol
DGlcD	Diglucosyl diacylglycerol
DHexD	Dihexosyl diacylglycerol
DNA	Deoxyribonucleic acid
Glc	Glucose
MGD	Monogalactosyl diacylglycerol
MGlcD	Monoglucosyl diacylglycerol
MHexD	Monohexosyl diacylglycerol
PAGE	Polyacrylamide gel electrophoresis
PG	Phosphatidylglycerol
SDS	Sodium dodecyl sulfate
TeGlcD	Tetraglucosyl diacylglycerol
TGlcD	Triglucosyl diacylglycerol

THexD
TeHexD
PL1
PL2

Trihexosyl diacylglycerol
Tetrahexosyl diacylglycerol
Phospholipid 1
Phospholipid 2

Nucleotide Sequence

B. subtilis ypfP

ttgaatacca ataaaagagt attaattttg actgcaaatt acggaaatgg acatgtgcag gtagccaaaa cactttatga
acaatgtgta cggctcggct ttcagcatgt aacagtttct aattgtacc aagagtcaaa tccgattgtt tcagaggtaa
ctcaatacct ttatttaaaa agctttctcaa tcgggaaaca gttttatcgt ttgttttatt acggagtga caaaatctat aataaacgta
aattcaatat ttactttaaa atgggtaata aaagattggg cgaactgtc gatgaacatc agcccgatat tattattaat
acatttccga tgatcgtcgt gccggaatac agacgccgaa ctggaagagt cattcctacc ttcaacgta tgactgatt
ttgtcttcat aaaatttggg ttcacgaaaa cgtggataaa tattatgtgg cgacagatta cgtgaaggaa aaactgctgg
agatcggcac tcatccaagc aatgtaaaaa tcacaggaat tccaatcagg ccgcaatttg aagaatccat gcctgttggc
ccgatatata aaaagtacaa tctttacca aacaaaaaag tgcttctgat catggcaggt gtcacgggtg tattaaagaa
cgtaaaagag ctgtgcgaaa acctgtcaa gtagaccaa gtgcaagtag ttgtcgtgtg cgggaaaaat acggctttaa
aagaatcttt gagtgcgctt gaagcggaaa atggtgacaa attaaaagtt ctgggctatg tggagcgcac tgatgagcta
tttggatca cagattgcat gattaccaag cccggcggca ttactttgac agaagccaca gccattggag tgcctgtcat
tctgtacaaa cccgtgcctg gccaggaaaa agaaaatgca aacttcttg aagaccgagg agctgccatc gttgtgaacc
gtcatgaaga gattctcgag tcagtcactt cccttcttg agatgaagat accttgcac gcataagaa aacattaag
gaccttcat tagcaaactc ctctgaagt atttagagg atactctgaa ggaatcagaa atgatgaccg ccaaacaaaa
agccaaagt ctatcgtaa

S. aureus ypfP

Atggttactca aaataaaaag atattgatta ttactggctc attcggtaac ggcatatgcaagttacaca gattatcgtt
aatcaactta atgatatgaa tctagaccat ttaagcgtcattgagcacga ttatttatg gaagtcac caattttgac
ttctattgt aaaaaatggt atatcaatag cttaaatat ttagaaata gtacaaagg gttttattac agccgcccag
ataaactaga caaatgtttt tacaatact atggacttaa taagttaatt aatttattgataaaagaaaa gccagattta
atattattaa cgtttctac accagttag tcggtactaa ctgagcaatt taacattaat attccagttg ctacagtgt
gacagactat cgcttacata aaaactggat tacgccgtat tcaacaagat attatgtggc aacaaaagaa acgaaacaag
acttcataga ctaggttatt gatccttcaa cagttaaagt gacaggtatt cctattgata acaatttga aacgcctatt
aatcaaaagc agtggttaat agacaacaac ttagatccag ataagcaaac tattttaatg tcagctggtg catttgggtg

atctaaaggt ttgacacga tgattactga tatattagcg aaaagtgcaa atgcacaagt agttatgatt tgtgtaaga
gcaaagagct aaagcgttct ttaacagcta agtttaaatt aacgagaatg tatttgattc taggttatac caaacacatg
aatgaatgga tggcatcaag tcaacttatg attacgaaac ctggtggtat cacaataact gaaggtttcg cccgttgat
tccaatgatt ttctaaatc ctgcacctgg tcaagagctt gaaaatgcct ttactttga agaaaaaggt ttggtaaaa
cgctgatac tccag

Amino acid sequence

B. subtilis YpfP

MNTNKRVLIL TANYGNHGVQ VAKTLYEQCV RLGFQHVTVS NLYQESNPIV
SEVTQYLYLK SFSIGKQFYR LFYYGVDKIY NKRKFNIYFK MGNKRLGELV
DEHQPDIIIN TFPMIVVPEY RRRTGRVIPT FNVMTDFCLH KIWVHENVDK
YYVATDYVK EKLLEIGTHPS NVKITGIPIR PQFEESMPVG PIYKKYNLSP
NKKVLLIMAG AHGVLKNVKE LCENLVKDDQ VQVVVVCGKN TALKESLSAL
EAENGDKLKV LGYVERIDEL FRITDCMITK PGGITLTEAT AIGVPVILYK
PVPGQEKENA NFFEDRGA AI VVNRHEEILE SVTSLLADED TLHRMCKNIK
DLHLANSSEV ILEDILKESE MMTAKQKAKV LS

S. aureus YpfP

MVTQNKKILI ITGSFGNGHM QVTQSIVNQL NDMNLDHLSV IEHDLFMEAH
PILTSICKKW YINSFKYFRN MYKGFYYSRP DKLDKCFYKY YGLNKLINLL
IKEKPDLILL TFPTPVMSVL TEQFNINIPV ATVMTDYRLH KNWITPYSTR
YYVATKETKQ DFIDVGIDPS TVKVTGIPID NKFETPINQK QWLIDNNLDP
DKQTILMSAG AFGVSKGFD T MITDILAKSA NAQVVMICGK SKELKRSLTA
KFKLTRMYLI LGYTKHMNEW MASSQLMITK PGGITITEGF ARCIPMIFLN
PAPGQELENA FYFEEKGFGK IADTPEEAIK IVASLTNGNE QLTNMISTME
QDKIKYATQT ICRDLLDLIG HSSQPQEIYG KVPLYARFFV K

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FIGURES:

Figure 1. Expression of the bacterial processive glycosyltransferases results in the biosynthesis of glycolipids in *E. coli* transformants. Lipid extracts were separated by TLC in chloroform: methanol: H₂O (70:30:4). Total lipids were detected with ANS under UV and marked with pencil, glycolipids were detected with α -naphthol/H₂SO₄ and tentatively identified by co-chromatography with standards.

lane 1: *E. coli* BL21 (DE3), control

lane 2: *E. coli* BL21 (DE3) pEypfP24, expressing the *B. subtilis* gene

lane 3: *B. subtilis* lipid extract

lane 4: *E. coli* BL21 (DE3) pEsay24, expressing the *S. aureus* gene

lane 5: standards, MGD, DGlcD and TGlcD

Figure 2. Negative ion mode MALDI-RE-TOF mass spectra of PL1 (top) and PL2 (bottom). The dominating molecular species of PL1 (top, [M-H]⁺ at m/z = 909) contains palmitic (16:0) and vaccenic acid (18:) residues, whereas the prevailing species of PL2 (bottom, [M-H]⁺ at m/z=1541) contains two palmitic (16:0) one palmitoleic (16:1) and one stearic acid (18:0) residue. The other species are described in the text.

(A, B, C)
Figure 3. Partial ¹H-NMR spectra (600 MHz, CDCl₃, 300K) of per-O-acetylated di-(2), tri-(3), and tetraglucosyl diacylglycerol (4).

Figure 4. Part of a 2D ROESY (upper) and 2D COSY (lower) spectrum (600 MHz, CDCl₃, 300K) of per O-acetylated triglucosyl diacylglycerol 3. NOE cross peaks used to assign the inter-residual connectivities are indicated in the ROESY spectrum as well as cross-peaks in the COSY spectrum. The corresponding parts of the 1D ¹H-NMR spectrum are displayed along the axes.

Figure 5. Part of 2D ROESY (bottom) and 2D COSY(top) spectra (600 MHz, CDCl₃, 300K) of PL1_{Ac,Me}. The corresponding parts of the 1D ¹H-NMR spectrum are displayed along the axes. The ROESY spectrum shows the connectivity between the anomeric proton H-1^A of

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Glc^A and the methylene protons H-3a^{Gro2} and H-3b^{Gro2}. A mixing time of 250 ms was used in the ROESY experiment.

Figure 6. Part of a ¹H-NMR spectrum (600 MHz, CDCl₃, 300K) of MGlcD_{Ac}(a), PL1_{Ac,Me}(b) DGlcD_{Ac}(c), and PL2_{Ac,Me}(d). Only those signals are indicated which are split due to the chirality in the phosphate group when compairing a/b and c/d. Despite their distance to the phosphate group a paticularly large effect is seen for the anomeric protons of the disaccharide moeity of PL2. Spectra were apodized by Gaussian multiplication with LB -1.5 and GB 0.2 prior to Fourier transformation.

Figure 7. Proton-decoupled ³¹P-NMR spectra (242.9 MHz, CDCl₃, 300K) of PL1_{Ac,Me} and PL2_{Ac,Me}. Two different phosphate resonances for each pair of diastereomeric phospholipids (P, P') of **PL1** (top) and **PL2** (bottom) are indicative of the chiral phosphate group in both phospholipids.

Figure 8. Structures of MGlcD, DGlcD, TGlcD, TeGlcD. The numbers underlined are related to the numbers in the text.

Figure 9: Structures of the two diasteriomeric forms of PL1_{Ac,Me} and PL2_{Ac,Me}. The chirality in the phosphate group resulted from the transformation of PL1_{Ac} and PL2_{Ac} to their methyl phosphates.

Figure 10. In vitro determination of acceptor specificities of bacterial processive glycosyltransferase from *S. aureus*.

Membrane fractions of *E. coli* BL21 (DE3) pEsay24 were used for *in vitro* enzyme assays with different labeled substrates as described in the experimental section. The lipophilic reaction products were subjected to thin-layer chromatography with subsequent radioscanning or fluorescence-detection for NBD-labelled products.

A) Enzyme assays with NBD-ceramide (NBD-Cer) as acceptor

lane 1: *E. coli* BL21 (DE3) control

lane2-4: independent *E. coli* BL21 (DE3) pEsay24

lane 5: NBD-ceramide-standard

The product of highest polarity present in lane 1-4 is a degradation product of NBD-ceramide.

B) Enzyme assays with radiolabeled sugar donors or different radiolabeled lipophilic acceptors and cell-free extracts of *E. coli* BL21 (DE3) pEsay24(lane 1-7).

lane 1-3: +[¹⁴C]cholesterol

lane 4-6: +[¹⁴C]cholesterolglucoside

lane 7: +UDP-[¹⁴C]glucose

lane 8: standard, [¹⁴C]cholesterolglucoside

lane 9: standard [¹⁴C]cholesterol

The labelled product with higher polarity in lane 10 was also present in *E. coli* cells transformed with pUC18 and is therefore not resulting from processive glycosyltransferase activity. The structural assignments are tentative and based on chromatographic behaviour.

Figure 11. In vitro demonstration of glucosyltransferase processivity of the glycosyltransferases from *B. subtilis* and *S. aureus* expressed in *E. coli* BL21 (DE). Cell extracts of *B. subtilis* and *E. coli* BL21 (DE3) expressing processive glycosyltransferase from *B. subtilis* and *S. aureus* were used for in vitro enzyme assays with UDP-[¹⁴C]-glucose. Internal DAG served as sugar acceptor. The lipophilic reaction products were subjected to thin-layer chromatography with subsequent radioscanning. The products were identified by co-chromatography with unlabeled standards.

Figure 12. In vitro demonstration of glucosyltransferase processivity of the glycosyltransferases from *B. subtilis* and *S. aureus* expressed in *E. coli* BL21 (DE). Membrane fractions of *E. coli* BL21 (DE3) expressing processive glycosyltransferase from *B. subtilis* and *S. aureus* were used for in vitro enzyme assays with unlabeled UDP-glucose and different radiolabeled acceptors as described in the experimental section. The lipophilic reaction products were subjected to thin-layer chromatography with subsequent radioscanning. The products were identified by co-chromatography with unlabeled standards.

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